Catalyst for the Conversion of Carbon Monoxide

Inventor: Takeda et al Atty. Docket: 20020802-004B

What is claimed is:

1. A catalyst for carbon oxide methanation reactions for fuel cells comprising a metal capable of forming

a metal-carbonyl species on a support having a predetermined pore size of sufficient dimensions to allow the

pore to accommodate a fully carbonylated metal complex.

5 2. The catalyst of Claim 1 wherein the support is a crystalline alumino-silicate.

3. The catalyst of Claim 1 wherein the support is selected from the group consisting of a molecular sieve,

beta-zeolite, mordenite, faujasite, any other alumino-silicate with a regular lattice structure, alumina, titania,

ceria, zirconia and combinations thereof.

The catalyst of Claim 3 wherein the support is selected from the group consisting of a beta-zeolite,

mordenite, and faujasite.

5. The catalyst of Claim 1 wherein the metal is selected from the group consisting of ruthenium, rhodium,

platinum, palladium, rhenium, nickel, iron, cobalt, lead, tin, silver, iridium, gold, copper, manganese, zinc,

zirconium, molybdenum, other metals that form a metal-carbonyl species and combinations thereof.

6. The catalyst of Claim 5 wherein the metal is selected from the group consisting of ruthenium, rhodium

and nickel.

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7. The catalyst of Claim 6 wherein the metal is ruthenium.

8. The catalyst of Claim 1 further comprising an inert binder.

9. The catalyst of Claim 8 wherein the binder is selected from the group consisting of alumina, γ -Al₂O₃,

SiO₂, ZrO₂, TiO₂ or pseudo-boehmite.

10. The catalyst of Claim 1 wherein the metal is added to the support through impregnation, incipient

wetness method, immersion and spraying.

11. The catalyst of Claim 7 wherein the ruthenium is added to the support through impregnation.

12. The catalyst of Claim 4 wherein the support has a pore volume in the range of from about 0.3cm³/g to

about 1.0cm³/g.

13. The catalyst of Claim 12 wherein the metal is ruthenium impregnated on the support so as to deliver a

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concentration of from about 0.5 wt% Ru to about 4.5 wt% Ru, based on the total weight of the catalyst

including the ruthenium.

14. A catalyst for carbon oxide methanation reactions for fuel cells comprising a metal capable of forming

a metal-carbonyl species on a support having a pore volume in the range of from about 0.3cm³/g to about

 $1.0 \text{cm}^{3}/\text{g}$. 5

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The catalyst of Claim 14 wherein the support is selected from the group consisting of a crystalline 15.

alumino-silicate, a molecular sieve, beta-zeolite, mordenite, faujasite, any other alumino-silicate with a regular

lattice structure, alumina, titania, ceria, zirconia and combinations thereof.

16. The catalyst of Claim 14 wherein the metal is selected from the group consisting of ruthenium,

rhodium, platinum, palladium, rhenium, nickel, iron, cobalt, lead, tin, silver, iridium, gold, copper, manganese,

zinc, zirconium, molybdenum, other metals that form a metal-carbonyl species and combinations thereof.

17. The catalyst of Claim 14 further comprising an inert binder.

The catalyst of Claim 17 wherein the binder is selected from the group consisting of alumina, γ-Al₂O₃, 18.

SiO₂, ZrO₂, TiO₂ or pseudo-boehmite.

19. The catalyst of Claim 14 wherein the metal is ruthenium impregnated on the support so as to deliver a

concentration of from about 0.5 wt% Ru to about 4.5 wt% Ru, based on the total weight of the catalyst

including the ruthenium.

A catalyst for carbon oxide methanation reactions for fuel cells comprising a metal selected from the 20.

group consisting of ruthenium, rhodium, platinum, palladium, rhenium, nickel, iron, cobalt, lead, tin, silver,

iridium, gold, copper, manganese, zinc, zirconium, molybdenum, other metals that form a metal-carbonyl

species and combinations thereof on a support having a pore volume in the range of from about 0.3cm³/g to

about 1.0cm³/g, wherein the support is selected from the group consisting of a crystalline alumino-silicate, a

molecular sieve, beta-zeolite, mordenite, faujasite, any other alumino-silicate with a regular lattice structure,

alumina, titania, ceria, zirconia and combinations thereof.

21. The catalyst of Claim 20 further comprising a binder selected from the group consisting of alumina, γ-

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Al₂O₃, SiO₂, ZrO₂, TiO₂ or pseudo-boehmite.

22. The catalyst of Claim 20 wherein the metal is ruthenium impregnated on the support so as to deliver a

concentration of from about 0.5 wt% Ru to about 4.5 wt% Ru, based on the total weight of the catalyst

including the ruthenium.

5 23. A catalyst for carbon oxide methanation reactions for fuel cells comprising ruthenium impregnated on

the support so as to deliver a concentration of from about 0.5 wt% Ru to about 4.5 wt% Ru, based on the total

weight of the catalyst including the ruthenium, wherein the support is selected from the group consisting of a

beta-zeolite, mordenite and faujasite.

24. The catalyst of Claim 23 wherein the support has a pore diameter of greater than about 6.3 Å and a pore

volume in the range of from about 0.3cm³/g to about 1.0cm³/g.

25. The catalyst of Claim 23 wherein the catalyst further comprises the binder γ-Al₂O₃ at a loading of

about 20 wt%, including the weight of the binder.

26. A method for carbon oxide methanation reactions for fuel cells using a catalyst comprising a metal

capable of forming a metal-carbonyl species on a support having a predetermined pore size of sufficient

dimensions to allow the pore to accommodate a fully carbonylated metal complex, the method comprising

passing a mixture of gases over the catalyst in a reaction zone having a temperature below the temperature at

which the shift reaction occurs and above the temperature at which the selective methanation of carbon

monoxide occurs.

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